

The Challenges of Determining Reliable Thermophysical Properties Required for CO₂ Carbon Capture and Storage.

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One primary way to reduce the carbon footprint of our hydrocarbon energy usage to drive modern technology is to capture the “greenhouse gases” produced by car, cattle, and heavy industry. Carbon dioxide is usually identified as one of primary offenders contributing as a “greenhouse” gas to “global warming.” One of the more important processes to reduce the amount of carbon dioxide in the atmosphere is to collect the gas from the atmosphere and then to store the gas in salt domes or depleted reservoirs that are the result of consuming the original contents. The transportation of large volumes of CO₂ from collection to storage involves pipelining the carbon dioxide from the collection site to the point where it is injected subsurface.

This presentation will discuss the challenges related to the thermophysical properties required to design CCS (Carbon Capture and Storage) and CO₂ transport systems. Particular focus will be on CO₂+ H₂O fluid mixtures and the water content of the liquid phase CO₂ at temperatures from 223 to 305 K and pressures from 0.35 to 7.50 MPa. The appearance of a water phase at subsea conditions for flowing CO₂ leads to pipeline corrosion problems (hydrates and ice form at 135 kPa and -50°C forms hydrates) as wells as gas at these temperatures and pressures. The liquid phase densities for liquid CO₂ at 10 MPa thermophysical properties of CO₂, and its mixtures in the supercritical region. Every viable application will have to be in the vicinity of the CO₂ critical point, making differences within the experimental data and predictions from thermodynamic models including CO₂ important and will be highlighted and discussed. One can then determine the risks of the transportation of CO₂ and other material to the injection point.